

# Analysis of Shallow Groundwater Monitoring Data of a Platinum Mine, Bushveld Complex, South Africa – Implications for Post-Closure

Robert Hansen

Director, Centre for Mineral Biogeochemistry, University of the Free State, PO Box 339, Bloemfontein, South Africa, hansenr@ufs.ac.za

## Abstract

Elevated total nitrogen ( $N_T$ ) and sulfate ( $SO_4^{2-}$ ) concentrations in shallow groundwater on platinum mine sites are a ubiquitous phenomenon. This is considered from mostly a regulatory perspective to be a serious water contamination problem and mining companies are forced to implement expensive mitigation and post-closure remediation measures. This work shows that natural biogeochemical processes have the potential to reduce  $N_T$  and  $SO_4^{2-}$  concentrations to below regulatory guideline values. Rates of  $N_T$  and  $SO_4^{2-}$  decrease in a post-closure scenario have been calculated from existing monitoring data.

Ten shallow groundwater monitoring data points from a platinum mine on the Western Limb of the Bushveld Complex, South Africa, representing a maximum period from February 2005 to October 2020 and 280 observations, were analysed. In addition to the standard timeseries and statistical analysis, a new analysis method was tested on the data. The method is termed “Plume-and-Pulse” analysis and entails identifying  $N_T$  and  $SO_4^{2-}$  pulses within a pollution plume from a timeseries plot of monitoring data and using the measured concentrations together with the proportion of nitrogen species i.e., nitrate ( $NO_3^-$ ), nitrite ( $NO_2^-$ ) and ammonium ( $NH_4^+$ ), of individual pulses and the differential change between analysis points to calculate rates of change of  $N_T$  and  $SO_4^{2-}$  species. These rates are statistically analysed and a distribution of post-closure  $N_T$  and  $SO_4^{2-}$  decrease rates estimated.

The main findings are that  $N_T$  and  $SO_4^{2-}$  show decreasing concentrations in shallow groundwater boreholes associated with mineral waste facilities, (i.e. tailings and waste rock), that have been intermittently operated over the monitoring period. The periods of inactivity represent post-closure scenarios. The Plume-and-Pulse analysis indicated a  $N_T$  range of decrease rates between  $5.25 \times 10^{-10}$  (635  $\mu\text{g/L/d}$ ) and  $9.54 \times 10^{-14}$  M/s (1  $\mu\text{g/L/d}$ ), with a median of  $2.39 \times 10^{-12}$  M/s (3  $\mu\text{g/L/d}$ ). The sulfate decrease rates range from  $7.32 \times 10^{-10}$  (6 075  $\mu\text{g/L/d}$ ) to  $1.24 \times 10^{-12}$  M/s (10  $\mu\text{g/L/d}$ ), with a median of  $1.26 \times 10^{-11}$  M/s (105  $\mu\text{g/L/d}$ ). The decrease of  $N_T$  may be caused by gasification as well as microbial reduction of dissolved N-species. The reduction of  $SO_4^{2-}$  is most probably caused by microbial S-reducers and precipitation of metal sulfides in the anaerobic groundwater aquifer matrix.

The implication of this work is that post-closure remediation of  $N_T$  and  $SO_4^{2-}$  may not be necessary, if decrease rates are rapid enough. If decrease rates are not rapid enough, simple augmentation of shallow groundwater aquifers by the relevant microbial species may be a sufficient, long-term, post-closure mitigation.

**Keywords:** Mine Pollution, Post-Closure, Nitrogen, Sulfate, Plume-and-Pulse Method

## Introduction

Elevated concentrations of total nitrogen ( $N_T$ ), from ammonium nitrate explosives (Brooks *et al.*, 2019), and sulfate ( $SO_4^{2-}$ ), from oxidising

metal sulfides in water draining from precious metal mine waste facilities is ubiquitous. The sources and geochemical behaviour of these constituents in the mine reticulation system

during the mining operational phase are generally well understood. However, as more mines are reaching the end of their lifetimes and are closing, knowledge of the fate of  $N_T$  and  $SO_4^{2-}$  in the natural environment, particularly in surface water and groundwater, is becoming increasingly important. From a biogeochemical perspective, the processes controlling the speciation and mobility in the impacted environment are less well understood.

With the advent of biogeochemistry as a scientific discipline from geochemistry by Vladimir Vernadsky and co-workers in the late 1990s (Schlesinger and Bernhardt, 2020), came the realisation that geochemical processes are more complex than previously thought. Life, especially microbial (e.g., bacterial), and algal species, exert major controls on the rate of geochemical processes in natural and anthropogenically impacted environments (Schlesinger and Bernhardt, 2020). This is especially true for essential bio-nutrients such as  $N_T$  and  $SO_4^{2-}$ .

In South Africa, mining companies are required to gather surface water and groundwater monitoring data for key parameters at key locations on the mine site during their operation. This data is sent to the relevant government departments and is meant to ensure compliance with conditions set out in the mine's water use licence. Often, this is the only purpose for which the monitoring data are used. In addition to compliance with regulations, monitoring data are also a record of biogeochemical processes and changes occurring at the mine site as well as in the surrounding environment. With sufficient knowledge, the monitoring data can be placed in a broader biogeochemical context, revealing the fundamental governing processes on specific constituents. Once the governing processes are known, their long-term behaviour can be determined and, if required, the appropriate remedial methods applied.

The aim of this study was to bring the continuous groundwater monitoring data at a Bushveld platinum mine, South Africa, into a biogeochemical context to identify impacts and whether remedial action in a post-closure scenario is required.

## Methods

Timeseries data from 10 groundwater monitoring points representing a maximum monitoring period of just over 15 years and 280 observations were provided by an operational platinum mine located in the Bushveld Complex, South Africa. The  $N_T$  and  $SO_4^{2-}$  data were analysed. The database contained data for three N species,  $NO_3^-$ ,  $NO_2^-$  and  $NH_4^+$ .

The monitoring points were selected downgradient of mine mineral waste facilities (i.e., waste rock and tailings) and downgradient of historically mined open pits and shafts. The data was analysed on three scales. Rates of change were calculated between several points of increasing and decreasing  $N_T$  and  $SO_4^{2-}$  concentration for each pulse. The individual pulse data was compiled for each monitoring point as well as for the entire mine site and analysed. Descriptive statistics, histograms and box plots were used as part of the data analysis process.

## Results and Discussion

Pulses of N were identified using the timeseries monitoring data. An example is fig. 1 which shows a borehole downgradient from a tailings facility. It is evident from the plot that the dominant N species is  $NH_4^+$  and that there appears to be an inverse correlation between  $NH_4^+$  and  $SO_4^{2-}$ . The interpretation of the timeseries monitoring data can be aided by the conceptual model in fig. 2. In fig. 2 the plume is shown as it is usually depicted in textbooks and publications. However, the plot at the bottom of fig. 2, which represents the monitoring data signal as measured over the monitoring period, and the actual data plot in fig. 1, shows that elevated N concentrations reach the monitoring borehole as pulses and is not a constant, continuous, homogenous input of N, and  $SO_4^{2-}$ , into the shallow groundwater system. The implication of the pulses of elevated  $N_T$  and  $SO_4^{2-}$  concentrations rather than a constant continuous input is that data from the analysis of individual pulses provide information on longer term trends of increasing or decreasing  $N_T$  and  $SO_4^{2-}$ . Analysis of a single pulse cannot provide information on the longer-term



biogeochemical behaviour of the system. The pulse-nature of the  $N_T$  and  $SO_4^{2-}$  plumes dictated that multiple pulses from multiple monitoring boreholes be analysed.

A total of 381 differential analysis points for  $N_T$  and 182 for  $SO_4^{2-}$  from 24 pulses in the 10 monitoring boreholes were analysed. The results of the analysis is shown in tab. 1 and graphically in fig. 3.

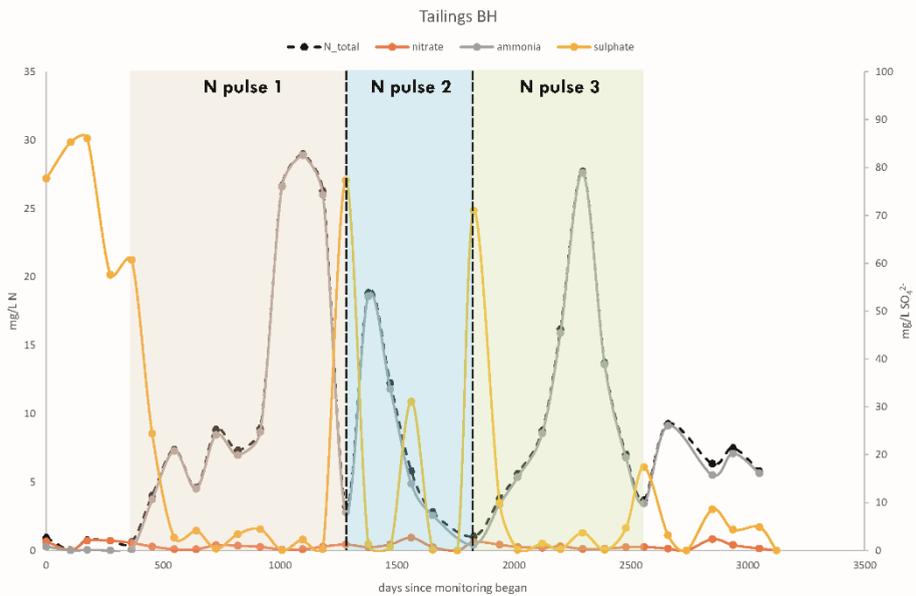


Figure 1 Timeseries plot of N and  $SO_4^{2-}$  data from a shallow groundwater monitoring borehole downgradient from a tailings facility showing the three N pulses analysed for differential changes in N species concentration.

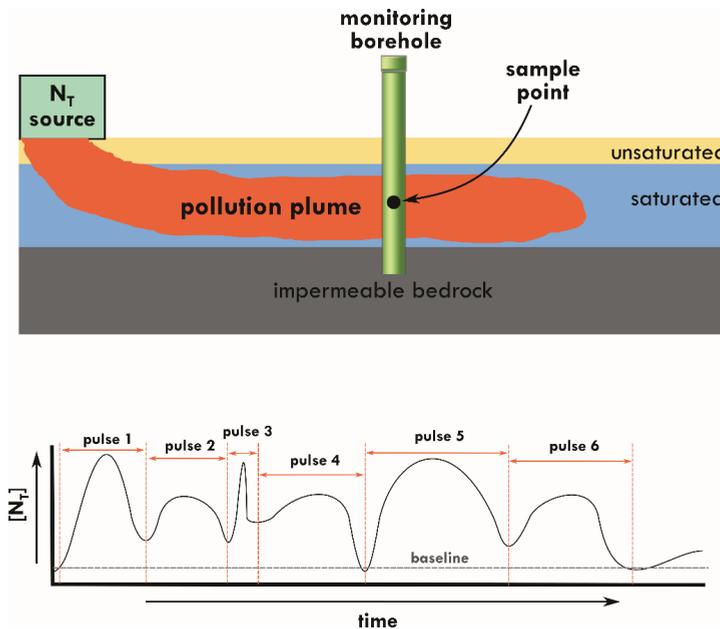


Figure 2 Conceptual model indicating the relationship between the signal in the timeseries monitoring data plot (bottom of the figure) and the plume emanating from a N-source e.g., tailings facility (top of figure).

*Table 1 Summary of concentration rates of change calculated from pulses in shallow groundwater monitoring borehole timeseries data.*

borehole (downgradient)	pulse	$\Delta[\text{SO}_4^{2-}]$		$\Delta[\text{N}_T]$	
		increase	decrease	increase	decrease
M/s					
Tailings BH 1	1	$2.61 \times 10^{-11}$			$-3.60 \times 10^{-12}$
	2		$-1.26 \times 10^{-11}$		$-1.19 \times 10^{-13}$
	3		$-4.05 \times 10^{-14}$		$-1.79 \times 10^{-12}$
Tailings BH 2	1		$-6.43 \times 10^{-11}$	$3.61 \times 10^{-12}$	
	2	$4.73 \times 10^{-11}$			$-2.39 \times 10^{-12}$
	3	$5.59 \times 10^{-11}$			$-1.02 \times 10^{-12}$
Tailings BH 3	1		$-8.71 \times 10^{-12}$		$-1.00 \times 10^{-13}$
	2		$-2.44 \times 10^{-12}$	$1.39 \times 10^{-13}$	
	3	$7.86 \times 10^{-12}$		$2.32 \times 10^{-12}$	
Tailings BH 4	1	$8.81 \times 10^{-12}$		$5.99 \times 10^{-13}$	
	2	$4.28 \times 10^{-12}$		$1.47 \times 10^{-13}$	
	3		$-1.04 \times 10^{-11}$		$-8.83 \times 10^{-14}$
Tailings BH 5	1	$2.19 \times 10^{-12}$		$7.60 \times 10^{-12}$	
	2	$8.82 \times 10^{-12}$		$2.59 \times 10^{-13}$	
	3		$-7.88 \times 10^{-12}$	$-1.88 \times 10^{-11}$	
Waste Rock BH 1	1	$5.51 \times 10^{-12}$			$-6.50 \times 10^{-13}$
	2		$-1.01 \times 10^{-9}$	$1.28 \times 10^{-10}$	
	3		$-1.09 \times 10^{-10}$		$-6.26 \times 10^{-10}$
Tailings BH 6	1	$1.18 \times 10^{-10}$			$-2.09 \times 10^{-10}$
	2	$1.02 \times 10^{-9}$		0	
	3		$-4.57 \times 10^{-10}$		$-4.57 \times 10^{-10}$
Waste Rock BH 2	1			$2.06 \times 10^{-12}$	
	2	$2.02 \times 10^{-10}$			
	3	$1.79 \times 10^{-10}$			$-5.20 \times 10^{-12}$
Waste Rock BH 3	1		$-3.08 \times 10^{-10}$		$-5.84 \times 10^{-12}$
	2				

The histograms and box plots of the  $\text{N}_T$  and  $\text{SO}_4^{2-}$  rate of change data indicates that the variation of the rates is within the same order of magnitude and that there are few outliers. It also shows that although the rates of increase and decrease of  $\text{N}_T$  and  $\text{SO}_4^{2-}$  overlap, the rates of decrease tend to be higher than the rates of increase, indicating a general trend towards decreasing concentrations of  $\text{N}_T$  and  $\text{SO}_4^{2-}$ .  $\text{N}_T$  decrease rates are between  $5.25 \times 10^{-10}$  (635  $\mu\text{g/L/d}$ ) and  $9.54 \times 10^{-14}$  M/s (1  $\mu\text{g/L/d}$ ), with a median of  $2.39 \times 10^{-12}$  M/s (3  $\mu\text{g/L/d}$ ). The sulfate decrease rates range

from  $7.32 \times 10^{-10}$  (6 075  $\mu\text{g/L/d}$ ) to  $1.24 \times 10^{-12}$  M/s (10  $\mu\text{g/L/d}$ ), with a median of  $1.26 \times 10^{-11}$  M/s (105  $\mu\text{g/L/d}$ ). The similarity between the order of magnitude of the rates of the increasing and decreasing concentrations indicate that the  $\text{N}_T$  and  $\text{SO}_4^{2-}$  redox processes may be coupled by biogeochemical processes.

Possible processes include volatilisation through denitrification forming predominantly  $\text{N}_2$  and  $\text{N}_2\text{O}$  gasses (Pina-Ochoa and Alvarez-Cobelas 2006, Korom 1992, Mateju *et al.* 1992) and the formation of ammonia through ammonification (Burger

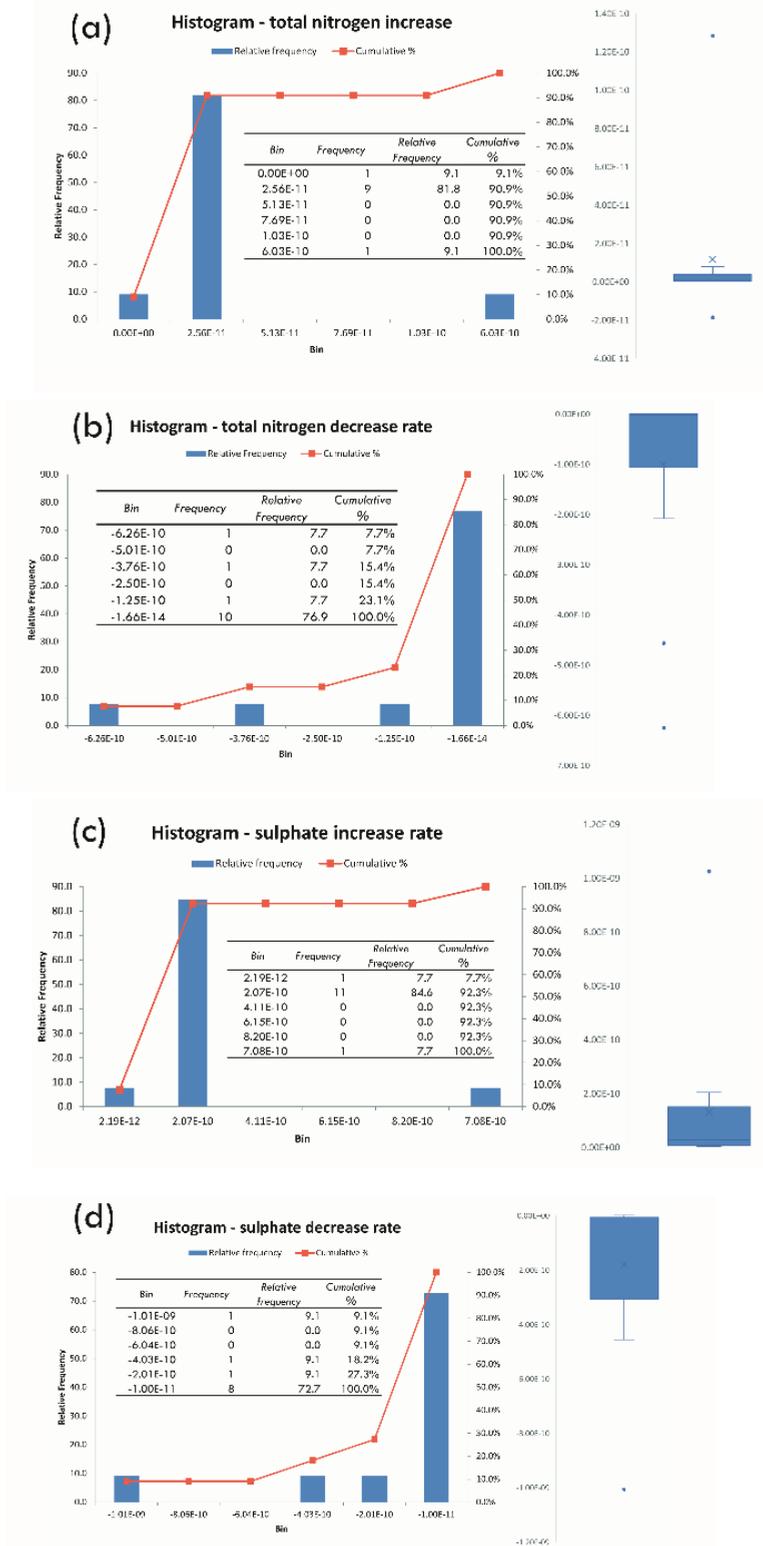
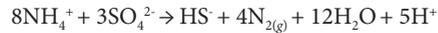


Figure 3 Histogram and box plots of the rates of change data for  $N_T$  (a and b) and  $SO_4^{2-}$  (c and d).

and Jackson 2003). The Henry constant for  $\text{NH}_3(\text{g})$  at 25°C per mole of gas dissolved is low (Rong *et al.* 2011) but can still be significant when considering the total mass of  $\text{N}_T$  in the system. Both processes can be microbially controlled and catalysed.

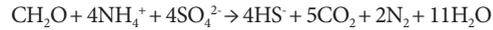
A potential mechanism by which  $\text{SO}_4^{2-}$  may be removed from at least the aquifer water column is reduction to sulfide ( $\text{S}^{2-}$ ), which at pH values greater than 7 is in the form of the  $\text{HS}^-$  anion. The  $\text{HS}^-$  then reacts with dissolved metal ions, principally iron ( $\text{Fe}^{2+}$ ) to form  $\text{S}^{2-}$  mineral precipitates e.g., pyrite [ $\text{FeS}_2$ ], which settle and become part of the aquifer matrix. This process can also be microbially controlled and catalysed (Shen and Buick, 2004).

The denitrification, ammonification, and sulphate reduction process can also be coupled by the microbial sulfamnox process. In recent years, laboratory studies on N and S biogeochemical processes show that N and S cycles can be linked in anaerobic environments found in marine and wetland sediments, by a  $\text{SO}_4^{2-}$ -reducing  $\text{NH}_3(\text{aq})^-$ -oxidation (sulfamnox) process (Liu *et al.*, 2021). The sulfamnox process is facilitated by microbes and entails the coupled reduction of  $\text{NH}_4^+$  and  $\text{SO}_4^{2-}$  to produce  $\text{N}_2(\text{g})$  and  $\text{N}_2\text{O}(\text{g})$ .  $\text{NO}_2^-$ ,  $\text{N}_2\text{H}_4$ ,  $\text{NH}_2\text{OH}$  and  $\text{S}_2^-$  are important intermediate species in this process. The net reaction for this process is given in eq. 1 (Liu *et al.*, 2021):



### Equation 1

In eq. 1, g denotes the gas phase. In the presence of organic matter, eq. 2 can be written (Liu *et al.*, 2021):



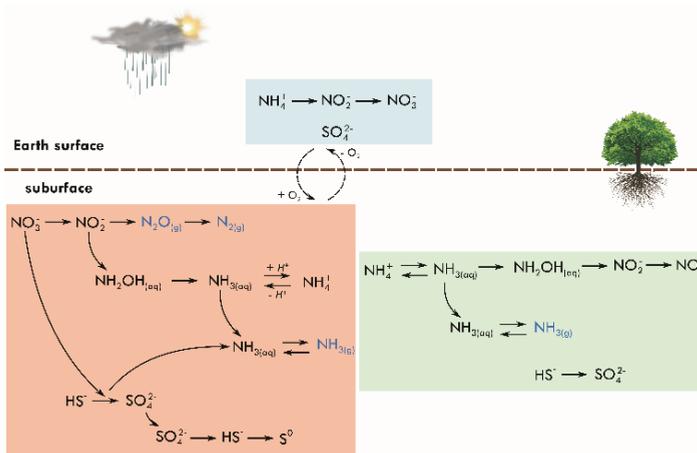
### Equation 2

The sulfamnox process occurs between pH values of 7 and 8.5 and at  $\text{HCO}_3^-$  concentrations < 1 000 mg/L (Liu *et al.*, 2021). A conceptual model of the process is shown in fig. 4.

Future work will be focussed on quantifying the biogeochemical processes responsible for the decrease of  $\text{N}_T$  and  $\text{SO}_4^{2-}$  in the platinum mine site groundwater system.

## Conclusions

An analysis of the rates of change of  $\text{N}_T$  and  $\text{SO}_4^{2-}$  from shallow groundwater monitoring data from boreholes downgradient of mine mineral waste facilities, shows a trend towards decreasing concentrations over time, implying that mass of these constituents is removed from the groundwater column to either the gas phase or to the solid phase (i.e., the aquifer matrix).  $\text{N}_T$  decrease rates are between  $5.25 \times 10^{-10}$  (635  $\mu\text{g/L/d}$ ) and  $9.54 \times 10^{-14}$  M/s (1  $\mu\text{g/L/d}$ ), with a median of  $2.39 \times 10^{-12}$  M/s (3  $\mu\text{g/L/d}$ ). The sulfate



**Figure 4** Reaction network diagram showing the sulfamnox process on the left, which couples nitrogen and sulfur reduction and the ammonification and sulfur oxidation process on the right. The balance within the network determines whether the concentration of N and S species decrease or increase over time.



decrease rates range from  $7.32 \times 10^{-10}$  (6 075  $\mu\text{g/L/d}$ ) to  $1.24 \times 10^{-12}$  M/s (10  $\mu\text{g/L/d}$ ), with a median of  $1.26 \times 10^{-11}$  M/s (105  $\mu\text{g/L/d}$ ).

The similarity between the order of magnitude of the rates of the increasing and decreasing concentrations indicate that the  $\text{N}_\text{T}$  and  $\text{SO}_4^{2-}$  redox processes may be coupled by biogeochemical processes.

Possible biogeochemical processes include denitrification forming  $\text{N}_2$  and  $\text{N}_2\text{O}$  gas as product, ammonification, which causes degasification of  $\text{N}_\text{T}$  in the form of  $\text{NH}_{3(\text{g})}$  as well as the reduction of  $\text{SO}_4^{2-}$  to  $\text{S}^2$  and the precipitation of metal sulfide minerals in the aquifer matrix. The denitrification and sulfate reduction may be linked by the sulfamox biogeochemical process. The implication of these results is that S and N can be expected to decrease in a post-closure scenario due to "natural" mitigation.

Future work will be focussed on quantifying the biogeochemical processes responsible for the decrease of  $\text{N}_\text{T}$  and  $\text{SO}_4^{2-}$  in the platinum mine site shallow groundwater system and the rates of decrease, focussed on a post-closure scenario.

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