

The release of dissolved inorganic carbon (DIC) and CO₂ from coal mine drainages

Dorothy J Vesper¹, Charles A Cravotta III², Kyle Fredrick³, Ellen K Herman⁴, Lili Lei⁵, Jill L Riddell⁶, Mathew L Bell¹, Lauren J Rockwell¹, Camille R Schaffer⁷

¹West Virginia University, United States of America;
²US Geological Survey;
³Penn West University, United States of America;
⁴Bucknell University, United States of America;
⁵Sweet Brier College, United States of America;
⁶Chatham College, United States of America;
⁷University Of Pittsburgh, United States of America

Extended Abstract

Coal utilization contributes to the release of geologically-bound carbon directly through combustion plus indirectly by accelerated weathering of carbonate minerals. Although CO_2 from combustion is instantly added to the modern carbon cycle, sulfuric-acid driven weathering of carbonate minerals tends to release geologically-bound carbon for decades. Coal mine drainage (CMD) commonly contains elevated concentrations of dissolved inorganic carbon (DIC) derived from weathering of carbonate overburden minerals, with corresponding partial pressure of CO_2 ranging from 10 to 1000 times greater than atmospheric equilibrium. Depending on the pH, the release of DIC can be in the form of CO_2 , which degasses to the atmosphere, or as dissolved bicarbonate which is exported downstream. The sulfuric-acid driven weathering of carbonate rocks in coal may have important implications for global carbon cycles because the old carbon is released without the concomitant drawdown of atmospheric carbon as with carbonic-acid driven weathering.

Our work employs the Anton Paar Carbonation Meter to obtain accurate and precise concentrations of CO₂ and DIC in water. These methods are modified from similar methods using the same instrument (Vesper and Edenborn 2012; Vesper et al. 2015). By obtaining data via this method, we avoid the pitfalls from determining CO₂ via back-calculation from either alkalinity or conventional DIC analysis. The back-calculation underestimates CO₂ pressures by excluding waters with low pHs (and hence no alkalinity). Standard methods of DIC analysis do not prevent the loss of CO₂ during analysis and therefore underestimate DIC in high-pCO₂ samples. Our approach allows us to determine the CO₂ concentration in low-pH and high-pCO₂ waters rather than excluding those hard-to-measure samples.

We have collected DIC and CO_2 data from CMD sites in the northern Appalachian Basin, including mines in West Virginia and from both the bituminous and anthracite regions of Pennsylvania (Vesper et al. 2016). The DIC concentrations almost always exceed atmospheric equilibrium values. Based on published and measured data, the release of DIC from individual mine discharges exceeds estimates and reported concentrations for carbonate springs or most terrestrial surface waters. Furthermore, DIC at a given CMD source can be relatively consistent or vary substantially over various time scales. Diurnal cycles, rain events, shifts in recharge, and seasonal changes in precipitation can alter the DIC concentration and flux (Bell 2020; Riddell 2015).

The implications of this work have yet to be fully determined although one study indicates that 140 mines in Pennsylvania have a comparable emission to a

small power plant (Vesper et al. 2016). Given that this is only a subset of the mine discharges in Pennsylvania, this emission is an underestimate of the total flux. Better regional-scale estimates will require more watershed-based spatial data and more precise discharge location information; additional sampling would help constrain possible temporal variations.

Keywords: CO₂, degassing, mine drainage

References

- Bell ML (2020) Relating recharge mechanisms to chemical changes in an updip Appalachian coal mine discharge: A case study from Lambert Run, West Virginia. In: Dept. of Geology and Geography. West Virginia University, Morgantown, WV, p 130.
- Riddell JL (2015) Comparing diel cycles of dissolved inorganic carbon to diel cycles of Fe and Mn at a coal mine drainage site in Harrison Co., WV. In: Dept. of Geology and Geography. West Virginia University, Morgantown, WV, p 138.
- Vesper DJ, Edenborn HM (2012) Determination of free CO2 in emergent groundwaters using

a commercial beverage carbonation meter. J Hydro 438–439(0):148-155. https://doi. org/10.1016/j.jhydrol.2012.03.015

- Vesper DJ, Edenborn HM, Billings AA, Moore JE (2015) A Field-Based Method for Determination of Dissolved Inorganic Carbon in Water Based on CO2 and Carbonate Equilibria. Water, Air, & Soil Pollution 226:28. https://doi.org/10.1007/ s11270-015-2348-z
- Vesper DJ, Moore JE, Adams JP (2016) Inorganic carbon dynamics and CO2 flux associated with coal-mine drainage sites in Blythedale PA and Lambert WV, USA. Environ Earth Sci 75(4):340. https://doi.org/10.1007/s12665-015-5191-z