

From Novel Laboratory Methodologies to Field Implementation: Assessing CO₂ and O₂ Flux in Northern Europe Mine Waste

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Abstract

The demand for metals in net-zero technologies has increased mining waste. This study examines carbon sequestration in ultramafic mine waste in Northern Europe. In collaboration with research partners, innovative methods assess CO₂ sequestration and emissions. Closed-system experiments showed O₂ depletion from 21% to negligible values (mg/L) due to sulfide oxidation. CO₂ removal is evaluated by monitoring atmospheric CO₂ and carbonate formation, considering CO₂ from acid mine drainage neutralisation. Results suggest carbonation can reduce CO₂ emissions, informing scalable carbon management. Open-barrel tests simulate field conditions, supporting sequestration strategies and contributing to the EU-funded C-SINK project.

Keywords: Gas flux monitoring, Carbon sequestration, Mine waste, Carbonation, Field trials

Introduction

This study expands on prior carbon capture and sequestration (CCS) research by Mine Environmental Management (MEM) Ltd. and Geochemic Ltd., focusing on CO₂ removal through mine waste materials. The increasing demand for metals essential to net-zero technologies has intensified mining activities, leading to significant waste production. Mine waste, particularly from ultramafic deposits, has the potential to sequester atmospheric CO₂ through mineral carbonation. However, challenges remain in quantifying CO₂ removal, understanding O₂ flux changes due to sulfide oxidation, and developing scalable carbon management strategies. Building on previous studies (e.g., Savage, 2019, 2023; Shiimi, 2022; Schoen, 2022), this study further investigates CO₂ sequestration potential using laboratory-scale and mesocosm experiments. The research integrates enhanced weathering (EW) techniques with improved monitoring,

reporting, and verification (MRV) frameworks. Additionally, preparations for field-scale testing aim to refine carbon dioxide removal (CDR) strategies in mining contexts, bridging the gap between controlled experiments and real-world applications.

Methods

Sample Preparation, Collection and Characterisation

Mine waste samples, obtained from waste rock storage facilities (WRSF) and tailings storage facilities (TSF) at Northern European mine sites, were characterised according to British (BS) and European (EU) standards. Characterisation focused on mineralogical (SEM-EDX, optical microscopy), elemental (ED-XRF for elemental distribution and carbon capture potential), geochemical (NAG, ANC, total carbon and sulfur, pH, EC), and geotechnical (specific gravity, particle size

distribution, water content) properties. These analyses identified oxidation and carbonation potential, assessed carbon sequestration capacity, and informed experimental design. Tailings were tested in Xylem WTW OxiTop®-C/B (Oxitops and Carbitops) and open-barrel setups, while waste rock underwent similar experiments. Historical MSc and PhD data (e.g., Savage *et al.* 2019, Chmielarski 2020, Schoen 2022, Shiimi 2022, Savage 2023, Cole 2023) supported sample selection and methodology

Closed-System Experiments

Laboratory-scale closed experiments utilised established methodologies to evaluate carbon capture. Oxitops experiments, conducted over 30 days at 8–10 °C, followed the WTW Soil BOD manual (2010a, 2010b) to measure O₂ consumption and CO₂ produced using a NaOH solution. Carbitops experiments, also lasting 30 days, were performed at 5% CO₂ and 8–10 °C to assess CO₂ consumption under elevated conditions. Sealed barrel experiments, spanning over 365 days, employed bespoke vessels developed by Schoen *et al.* (2023), using controlled starting gas compositions (O₂, CO₂) at 8–10 °C. Throughout these experiments, pressure and gas concentration changes were monitored to evaluate CO₂ production and consumption rates, enabling detailed carbon balance calculations.

Open-System Experiments

An open barrel mesocosm test (Fig. 1) was designed to simulate field conditions, in Northern Europe, providing a platform to evaluate sensor performance before upscaling to field trials. The experiment operated under atmospheric O₂ and CO₂ conditions at 8–10 °C, with the objective of developing and setting up an open-barrel equipment test to monitor gas flux and geochemical behavior in mine tailings under semi-controlled open system conditions. This development represents a new approach to understanding carbonation processes in mining contexts. As part of a broader EU-funded carbon sequestration project, open system experimentation is considered part of the plans for an upscaled field trial at a Northern European mine site.

Results

Characterisation of ultramafic mine waste

Waste rock samples had a moisture content of 7.5% added before experiments, as water is necessary for the carbonation process. Carbon dioxide dissolves in water to form carbonic acid, which reacts with Mg- and Ca-bearing silicates to produce secondary carbonate minerals. Particle size distribution (PSD) curves for waste rock revealed well-graded material ranging from gravel to fines, with smaller particle sizes contributing increased surface area for oxidation and

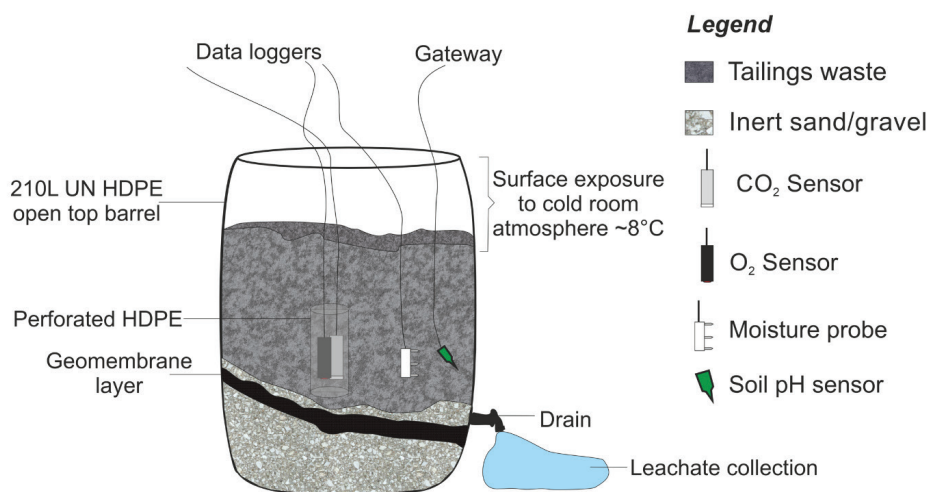


Figure 1 Open barrel schematic (designed and submitted as part of first author's MSc dissertation for Cardiff University).



carbonation reactions. Elemental analysis from prior studies was used to evaluate the carbon capture potential (CCP) for waste rock and tailings. Waste rock demonstrated a higher CCP (~ 370 kg CO₂/tonne) compared to tailings (~ 330 kg CO₂/tonne). Calcium concentrations remained consistent ($\sim 10\%$) across both materials. Magnesium content was higher in waste rock, potentially explaining its elevated CCP due to magnesium's role in carbon capture. The mineralogical compositions of waste rock and tailings were broadly similar featuring sulfides, silicates and carbonates. Target minerals such as pyrrhotite, pyrite, calcite, and dolomite were identified, alongside gangue minerals including diopside, enstatite, tremolite, serpentine, and chlorite. These minerals are critical for carbonation reactions, as sulfides generate acidity, carbonates neutralise it to produce CO₂, and magnesium silicates react with CO₂ to form carbonates.

Carbon Balance by Oxitops and Carbitops

Under low-temperature conditions (10 °C), the carbon balance of the analysed Oxitops and Carbitops mine waste samples demonstrated net CO₂ sequestration potential (Fig. 2). CO₂ production rates estimated from both the back titration and TIC methods were estimated between 0.40 and 1.24 kg CO₂/tonne/year for tailings and less than 0.10 kg CO₂/tonne/year for waste rock. In a 5% CO₂ atmosphere, CO₂ consumption rates ranged from 1.90 to 2.11 kg CO₂/tonne/year for tailings and 1.16 to 1.25 kg CO₂/tonne/year for waste rock. Net CO₂ sequestration rates were higher in waste rock, ranging from 1.09

to 1.20 kg CO₂/tonne/year, driven by limited CO₂ production from sulfide oxidation and carbonate dissolution, coupled with notable CO₂ uptake by silicate minerals. For tailings, net sequestration rates were estimated between 0.67 and 1.71 kg CO₂/tonne/year, with higher CO₂ production from sulfide oxidation offset by greater CO₂ uptake via carbonation reactions.

Passive Carbonation in Closed Barrels

Two sealed barrels (Fig. 3) stand out as they directly compare to the hypotheses for atmospheric conditions developed by Schoen *et al.* (2023). A waste rock barrel, exposed to atmospheric conditions, showed O₂ concentrations starting at $\sim 20\%$ (≈ 278 mg/L), in November 2022, which gradually decreased over time due to sulfide oxidation, reaching complete depletion by April 2024. This aligns with the hypothesis for atmospheric CO₂ conditions, which predicted that O₂ concentrations would fall as sulfides were oxidised, leading to a reduction in available O₂. CO₂ concentrations in this barrel began at ~ 400 ppm (≈ 0.56 mg/L) and steadily increased, peaking at $\sim 1.6\%$ ($\approx 16,000$ mg/L) in April 2024, coinciding with the complete O₂ depletion. This supports the hypothesis that CO₂ levels would initially increase due to the rapid occurrence of sulfide oxidation and primary carbonate dissolution, releasing CO₂ before passive carbonation (CO₂ uptake) becomes dominant. From mid-April 2024 onward, CO₂ concentrations began to decline, reaching $\sim 1.3\%$ ($\approx 13,000$ mg/L) by November 2024. A control barrel, which contained quartz chips, showed little change

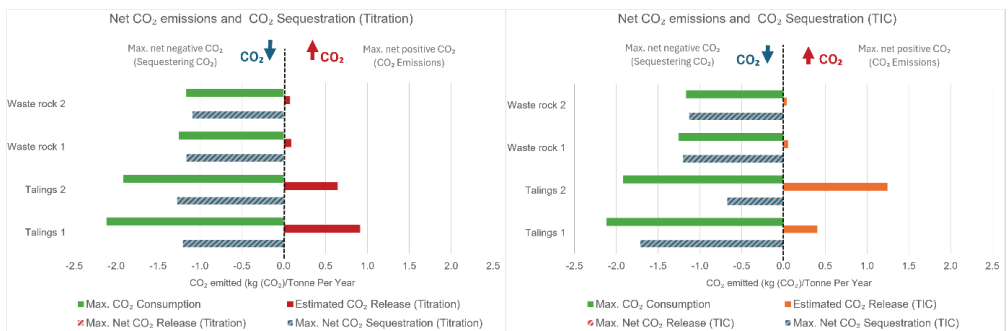


Figure 2 Net CO₂ emissions and sequestration results via titration and total inorganic carbon (TIC) methods.

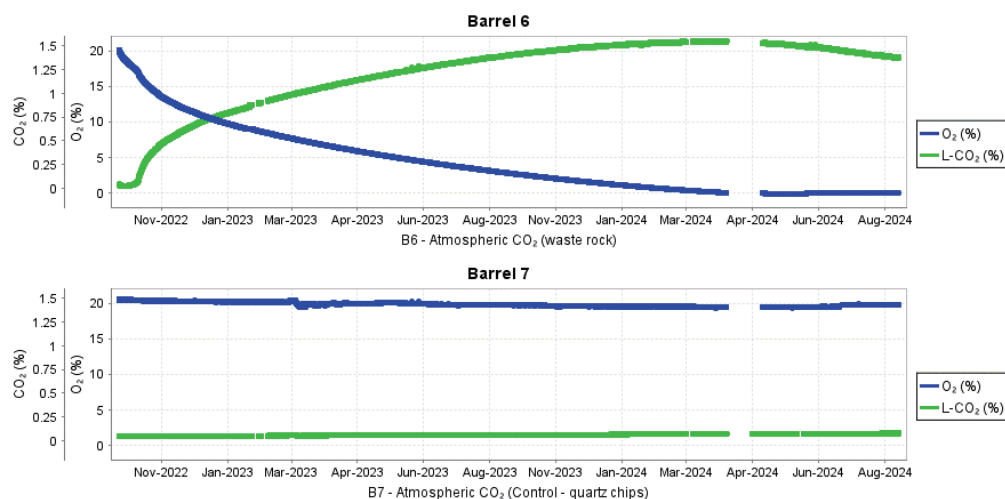


Figure 3 Sealed barrel results after prolonged monitoring of the gas flux.

in O₂ and CO₂ concentrations over the two-year period. A minor deflection in O₂ levels between November 2023 and May 2024 was attributed to sensor corrosion, confirming that no reaction-driven gas exchange occurred in this control barrel.

Commissioning the Open Barrel Equipment Test

The open barrel was commissioned and filled with mine tailings sourced from a Northern European mine site. The barrel is instrumented with O₂, CO₂, pH, temperature, and soil moisture sensors (Fig. 4). The barrel will be irrigated, and leachate collected for geochemical analysis. The open barrel contains 120 kg of homogenised mine tailings at 20% moisture content, with a base layer of 25 kg of DI-rinsed river aggregate gravel and 20 kg of inert sand mix. The system also includes a geomembrane to contain and prevent contamination of the materials. The open barrel mesocosm test will enable more precise measurements of the factors influencing carbonation processes, which is vital for optimising carbon sequestration strategies. Real-time monitoring results from this new approach will provide valuable insights into CO₂ removal efficiency in mining contexts, paving the way for more sustainable practices and effective carbon management across the industry.

Key Observations/Discussion

Critical characterisation of ultramafic mine waste

The characterisation of waste rock and tailings include the importance of moisture content, as water is essential for carbonation by enabling CO₂ dissolution and subsequent reactions with magnesium- and calcium-bearing silicates to form carbonates. The particle size distribution showed a range of material sizes, with smaller particles offering a larger surface area for oxidation and carbonation. Elemental and mineralogical analyses revealed similarities between the materials, with both containing mafic minerals like pyroxene and amphibole, though tailings had a higher presence of olivine. Various key minerals, including pyrrhotite, pyrite, calcite, and dolomite, were identified as important for carbonation processes. These characteristics suggest that both waste rock and tailings possess the necessary conditions for carbonation, with mineral composition playing a considerable role in their carbon sequestration potential.

Nature of the closed system experiments

The Oxitop and Carbitop tests demonstrated that both tailings and waste rock have the potential for CO₂ sequestration. Waste rock exhibited higher net CO₂ uptake due to lower CO₂ production from sulfide oxidation and



Figure 4 Open barrel equipment setup.

greater carbonation of silicate minerals. In contrast, tailings generated more CO₂ from sulfide oxidation but also underwent substantial carbonation, resulting in net sequestration. The saturated nature of the tailings likely restricted CO₂ diffusion, enhancing its uptake by silicate minerals. The largely saturated conditions in the tailing's facility likely limited CO₂ diffusion relative to O₂, allowing CO₂ produced during sulfide oxidation to be consumed by silicate minerals. These findings indicate that both waste rock and tailings can achieve net CO₂ sequestration under the studied conditions, with tailings exhibiting higher carbonation rates than waste rock.

The sealed barrel tests further supported these findings. The waste rock barrel followed expected trends, with O₂ depletion due to sulfide oxidation and an initial CO₂ increase from sulfide oxidation and carbonate dissolution. Over time, CO₂ concentrations declined as carbonation reactions became dominant. The control barrel, containing quartz chips, exhibited negligible gas fluctuations, confirming the absence of reaction-driven gas exchange.

The future of the open system experiments

While initial sensor readings for the open barrel equipment test are being recorded, the timeline for carbonation development remains uncertain, and the full trends will emerge as data collection continues. The future field trials will provide more

information to refine carbonation process insights and carbon sequestration strategies.

Conclusion

The project has made substantial contributions to the advancement of carbon sequestration strategies by utilizing mine waste materials. By building on prior research, this study has refined methodologies for monitoring CO₂ and O₂ fluxes, a critical aspect of assessing the carbon capture potential of mine waste. These efforts not only address Measurement, Reporting, and Verification (MRV) challenges but also enhance Enhanced Weathering (EW) frameworks for Carbon Dioxide Removal (CDR). The novel open-barrel experiment, along with upcoming field trials at the mine site, marks a crucial step in validating these strategies in real-world settings. The data gathered from these trials will offer valuable insights that will shape future carbon sequestration technologies, including those supported by the EU-funded C-SINK project. This work establishes a solid foundation for improving carbon sequestration methods and advancing efforts to mitigate atmospheric CO₂ levels. Future research should focus on modelling existing CCS data (e.g. using PHREEQC) to identify emerging trends and further refine conclusions. Additionally, enhancing O₂ and CO₂ consumption rate models using techniques such as differential or integrated rate laws will considerably improve carbon sequestration models. Investigating parameters like temperature,

O₂/CO₂ content, particle size, and mineralogy will be essential in developing more accurate and robust consumption estimation models for future carbon capture strategies.

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